Philadelphia University Faculty: Pharmacy

**Department: Pharmacy** 



Approved Date: 16/10/2022 Issue: 1

Academic Year: 2022-2023

**Course Syllabus** 

Credit Hours: 3 Bachler: 2<sup>nd</sup> year

#### **Course Information**

Course No.	Course Title	Prerequisite	
0510211		Pharmaceutical Organic	
0310211	Pharmaceutical Organic Chemistry II	Chemistry I (	0521211)
	Course Type	Class Time	Room No.
Univirsity Re	equirementFuclty Requirement	Sunday-Tuesday:	
MajorRequir	ement Elective Compulsory		
5 1	1	9:45-11:15 am	
		for Section-2	
		&	
		12:45- 02:15 pm	
		for	
		Section-1	

#### **Instructor's Information**

Name	Office No.	Phone No.	Office Hours	E-mail

### **Course Delivery Method**

Blended	Online Physic		cal		
Learning Model					
Demonstrage	Synchronous	Asynchronous	Physical		
Percentage			100%		

#### **Course Description**

This course is a continuation of pharmaceutical organic chemistry (I). It includes the study of the physical properties, nomenclature, preparation methods, identification methods and reactions of the cyclic and acyclic organic compounds: aromatic compounds, alcohols and phenols, ethers, thiols and sulfides, aldehydes, ketones, carboxylic acids and their derivatives and amines.

Course	Learning	Outcomes
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Number	Outcome	Corresponding Program Outcomes	Corresponding Competencies
	Knowledge		
K1	To acquire knowledge of naming different organic compounds	K <sub>P</sub> 1	C1
K2	To describe different organic reactions and their mechanisim	K <sub>P</sub> 1	C1
K3	To demonstrate the relative reactivity, acidity, and bacisty of organic compounds	K <sub>P</sub> 1	C1
	Skills		
S1	Drawing the chemical stracture of given organic molecules	S <sub>P</sub> 2	C8
S2	Able to arrange the molecules according to their reactivity, or acidity, or bacisty	Sp2	C8
\$3	Writing the reaction mechanisuim and Plan to synthesize different compounds according to their understanding of functional groups reactions.	Sp2	C8

# Learning Resources

Course Textbook	Organic Chemistry, John Mc-Murry, 2016, 9th edition.			
Supporting References	1. Organic Chemistry, SOLOMN and FRYHLE, Wily international			
	brooks/ cale, Thomson learning, 2011, 10th. Edition.			
	2.Study Guide with Student Solutions Manual for McMurry's Organic chemistry			
Supporting Websites	https://www.youtube.com/			
Teaching Environment	assroomlaboratoryLearnin Platform Other			

# Meetings and Subjects Time Table

Week	Торіс	Learning Method*	Task	Learning Material
1	<b>Benzene and aromaticity</b> Sources and names of aromatic compound, structure and stability of benzene, aromaticity and huckel rule, aromatic ions, aromatic heterocycles : pyridine and pyrrole, poly cyclic aromatic compounds	Lecture Flipped learning <b>Problem</b> solving based learning	homework	Text book Chap 15

	Chemistry of benzene	Lecture	homework	
	Electrophilic aromatic substitution			
	reactions, : Bromination and other			
	aromatic substitutions, alkylation and	Flipped		
	acylation Substituent effects in	learning		
2&3	substituited aromatic rings , an			Text book
	explanation of substituent effects,	Problem		Chap 16
	trisubstituted benzenes, nucleophilic	solving		
	aromatic substitutions, benzyen,	Dased		
	reduction of aromatic compounds,	learning		
	synthesis of trisubstituedbenzens			
	Alcohols and Phenols: structure and	Lecture	homework	
	nomenclature acidity of	Lecture	nomework	
	phenols synthesis reactions analysis of	Flipped		
	alcohols and phenols	learning		
4&5				Text book
		Problem		Chap 17
		solving		
		based		
		learning		
	Ethers and Epoxides	Lecture	homework	
	Names and properties of ethers, synthesis			
	of ether, reactions	Flipped		
-	Reaction of epoxides: ring opening	learning		Text book
6		<b>D</b> 11		Chap 18
		Problem		
		solving		
		learning		
	Aldehyde and ketones: structure and	Lecture	homework	
	nomenclature, preparations.	Lecture	nomework	
	reactionsincluding(oxidation, reduction	Flipped		
7.0	Cannizzaro reaction).	learning		
/,ð 8-0		C C		Lext DOOK
& 9		Problem		Chap 19
		solving		
		based		
		learning		
	Carboxylic acids and nitriles, structure	Lecture	homework	
	and nomenclature, Physical properties	T21' 1		
	acidity and their reactions.	Flipped		
10 &		learning		Text book
11		Duchlom		Chap 20
		Problem		
		based		
		learning		
	Carboxylic acid derivatives: Acid	Lecture	homework	
12&	chlorides, Acid anhydrides. Amides and			Text book
13	Esters, Their structures and	Flipped		Chap 21
	nomenclature, synthesis, reactions.	learning		-

		Problem solving based learning		
	Amines: nomenclature, basicity,	Lecture	homework	
	preparation, reaction, diazonium saits			
	and their reactions.	Flipped		
14&		learning		Text book
15		Problem		Chap 24
		solving		
		based		
		learning		
16	Final Exam			

\*Includes:lecture, flipped Class, project based learning, problemsolvingbased learning, collaboration learning.

## **Course Contributing to Learner Skill Development**

Using Technology			
Using chemdraw program			
Using power point or any related programs for preparing presentations			
Communication Skills			
Application of Concept Learnt			

## Assessment Methods and Grade Distribution

Assessment Methods	Grade	Assessment Time (Week No.)	Course Outcomes to be Assessed
Mid Term Exam	% 30	11 <sup>th</sup> week	K1, K2,K3
			51,52,53
Term Works*	% 30	Continuous	
<b>Final Exam</b>	% 40	16 <sup>th</sup> week	K1, K2,K3
			<b>S1,S2,S3</b>
Total	%100		

\* Include: quizzes, in-class and out of class assignment, presentations, reports, videotaped assignment, group or individual project.

Number	Learning Outcomes	Corresponding	Learning Mothod*	Assessment
	Knowledge	competences	Method*	Ivietnou**
K1	To acquire knowledge of naming	C1	Lecture	Exam/subjective
	different organic compounds		Lieture	and objective
			Flipped	questions
			learning	_
			Problem	
			solving	
			learning	
K2	To describe different organic	C1.	Lecture	Exam/subjective
112	reactions and their mechanisim	01,	Lecture	and objective
			Flipped	questions
			learning	1
			_	
			Problem	
			solving	
			based	
K2	To domonstrate the relative	C1	Learning	Exam/aubiaativa
KJ	reactivity acidity and bacisty	CI	Lecture	and objective
	of organic compounds		Flipped	questions
			learning	questions
			U	
			Problem	
			solving	
			based	
	Skille		learning	
<u>\$1</u>	Drawing the chemical stracture	C8	Lecture	Exam/subjective
51	of given organic molecules	0	Lecture	and objective
			Flipped	questions
			learning	-
			Problem	
			solving	
			based	
\$2	Able to arrange the molecules	C8	Lecture	Exam/subjective
04	according to their reactivity or	CO	Little	and objective
	acidity, or bacisty		Flipped	auestions
			learning	1
			U	
			Problem	
			solving	
			based	
		<u> </u>	learning	
85	writing the reaction	Cð	Lecture	Exam/subjective
	mechanisunn and Fian to			and objective

# Alignment of Course Outcomes with Learning and Assessment Methods

synthesize different compounds according to their understanding of functional groups reactions.	Flipped learning	questions
or runeuonai groups reactions.	Problem solving based learning	

\*Include: lecture, flipped class, project based learning, problem solving based learning, collaboration learning. \*\* Include: quizzes, in-class and out of class assignments, presentations, reports, videotaped assignments, group or individual projects.

### **Course Polices**

Policy	Policy Requirements			
Passing Grade	The minimum pass for the course is $(50\%)$ and the minimum final mark is $(35\%)$ .			
Missing Exams	<ul> <li>Anyone absent from a declared semester exam without a sick or compulsive excuse accepted by the dean of the college that proposes the course, a zero mark shall be placed on that exam and calculated in his final mark.</li> <li>Anyone absent from a declared semester exam with a sick or compulsive excuse accepted by the dean of the college that proposes the course must submit proof of his excuse within a week from the date of the excuse's disappearance, and in this case, the subject teacher must hold a compensation exam for the student.</li> <li>Anyone absent from a final exam with a sick excuse or a compulsive excuse accepted by the dean of the college that proposes the material must submit proof of his excuse within three days from the date of holding that exam.</li> </ul>			
Attendance	The student is not allowed to be absent more than (15%) of the total hours prescribed for the course, which equates to six lecture days (n t) and seven lectures (days). If the student misses more than (15%) of the total hours prescribed for the course without a satisfactory or compulsive excuse accepted by the dean of the faculty, he is prohibited from taking the final exam and his result in that subject is considered (zero), but if the absence is due to illness or a compulsive excuse accepted by the dean of the college that The article is introduced, it is considered withdrawn from that article, and the provisions of withdrawal shall apply to it			
Academic Integrity	Philadelphia University pays special attention to the issue of academic integrity, and the penalties stipulated in the university's instructions are applied to those who are proven to have committed an act that violates academic integrity, such as cheating, plagiarism (academic theft), collusion, intellectual property rights.			

### **Program Learning Outcomes to be Assessed in this Course**

Number	Learning Outcome	Course Title	Assessment Method	Targeted Performance level

## **Description of Program Learning Outcomes Assessment Method**

Number	Detailed Description of Assessment	

## Assessment Rubric of the Program Learning Outcomes